

**HONG KONG SPECIALTY GASES CO., LTD.**

2/F., Blk. B, Wing Cheong Fty. Bldg., 121 King Lam St.,

Cheung Sha Wan, Kln., Hong Kong.

**MATERIAL  
SAFETY  
DATA SHEET**

<b>PRODUCT NAME</b> Dichlorosilane	<b>CAS#</b> 4109-96-0
<b>TRADE NAME AND SYNONYMS</b> Dichlorosilane	<b>DOT I.D. NO.</b> UN 2189
<b>CHEMICAL NAME AND SYNONYMS</b> Dichlorosilane	<b>DOT HAZARD CLASS</b> Division 2.3
<b>ISSUE DATE AND REVISIONS</b> Revised November 2000	<b>FORMULA</b> SiH <sub>2</sub> Cl <sub>2</sub>

**HEALTH HAZARD DATA**

<b>EMERGENCY OVERVIEW</b> Dichlorosilane is a colorless, flammable, toxic gas at room temperature and atmospheric pressure. It has an irritating odor and it fumes in moist air, being hydrolyzed to hydrochloric acid and a mixture of polymeric siloxanes. It is shipped as a liquefied gas in low pressure steel cylinders at a pressure of 66 kPa (9.6 psig) at 20°C.
<b>SYMPTOMS OF EXPOSURE</b> Corrosive and irritating to the upper and lower respiratory tracts, skin and eyes. It hydrolyzes very rapidly in water or with the moisture in the air yielding hydrogen chloride. Skin burns and mucousal irritation are like that from exposure to hydrochloric acid. Symptoms include lacrymation, cough, labored breathing and excessive salivary and sputum formation. Excessive irritation of the lungs causes acute pneumonitis and pulmonary edema which could be fatal. Hydrochloric acid burns exhibit severe pain, redness, possible swelling and early necrosis.
<b>TOXICOLOGICAL PROPERTIES</b> Dichlorosilane is irritating and corrosive to all living tissues. Toxic level exposure to dermal tissue causes hydrochloric acid burns and skin lesions resulting in early necrosis and scarring. Chemical pneumonitis and pulmonary edema result from exposure to the lower respiratory tract and deep lung. Residual pulmonary malfunction might also occur. Burns to the eye result in lesions and possible loss of vision.
<b>RECOMMENDED FIRSTAID TREATMENT</b> <b>PROMPT MEDICAL ATTENTION IS MANDATORY IN ALL CASES OF OVEREXPOSURE TO DICHLOROSILANE. RESCUERS SHOULD BE EQUIPPED WITH ADEQUATE PERSONAL PROTECTIVE APPARATUS.</b>  <u>Inhalation:</u> Conscious persons should be assisted to an uncontaminated area and inhale fresh air. Unconscious persons should be moved to an uncontaminated area and given assisted respiration and supplemental oxygen. Keep the victim warm and quiet. Assure that mucus or vomited material does not obstruct the airway by positional drainage. Delayed pulmonary edema may occur. Keep patient under medical observation for at least 24 hours. <u>Eye Contact:</u> Persons with potential exposure to dichlorosilane should not wear contact lenses. Flush contaminated eye(s) with copious quantities of water. Part eyelids to assure complete flushing. Continue for at least 15 minutes. <u>Skin Contact:</u> All affected body surfaces be washed with copious quantities of running water for sufficient time. Remove contaminated clothing as rapidly as possible.

## HAZARDOUS MIXTURES OF OTHER LIQUIDS, SOLIDS, OR GASES

Reacts vigorously with water or moist air yielding hydrogen chloride. It is also flammable in air over a very wide range.

## PHYSICAL DATA

<b>BOILING POINT</b> 8.4°C (47.1°F)	<b>CRITICAL TEMPERATURE</b> 176.0°C
<b>MOLAR SPECIFIC HEAT</b> (25 oC, 1 bar abs, constant volume) 62.053J/mol <sup>o</sup> K	<b>CRITICAL PRESSURE</b> 46.76 bar abs
<b>SOLUBILITY IN WATER</b> Hydrolyzes	<b>SPECIFIC VOLUME</b> (@21.1 °C, 101.325 kPa) 239.1 dm <sup>3</sup> /kg
<b>EVAPORATION RATE</b> N/A (Gas)	<b>SPECIFIC GRAVITY (AIR=1)</b> @ 70°F (21.1°C) = 3.48
<b>APPEARANCE AND ODOR</b> Colorless, flammable, toxic gas with acidic, irritating odor	

## FIRE AND EXPLOSION HAZARD DATA

<b>FLASH POINT</b> (Method used) 62 °F (C.C.)	<b>AUTO IGNITION TEMPERATURE</b> 212 °F (100 °C)	<b>FLAMMABLE LIMITS % BY VOLUME</b> LEL 4.1 UEL 98.8
<b>EXTINGUISHING MEDIA</b> Carbon dioxide or dry chemical on small fires. Coarse water spray on large fires.		
<b>SPECIAL FIRE FIGHTING PROCEDURES</b> A coarse water spray should be used on large fires since the hydrolysis reaction to HCl is more rapid than the combustion reaction.		
<b>UNUSUAL FIRE AND EXPLOSION HAZARDS</b> Recognize the potential for ground (water) contamination with hydrochloric acid following a dichlorosilane fire which has been "extinguished" with water.		

## REACTIVITY DATA

<b>STABILITY</b>		<b>CONDITIONS TO AVOID</b> None
Unstable		
Stable	X	
<b>INCOMPATIBILITY</b> (Materials to avoid) Water, oxidizers		
<b>HAZARDOUS DECOMPOSITION PRODUCTS</b> Hydrogen chloride, hydrogen, chlorine and finely divided amorphous silica		
<b>HAZARDOUS POLYMERIZATION</b>		<b>CONDITIONS TO AVOID</b> None
May Occur		
Will Not Occur	X	

## SPILL OR LEAK PROCEDURES

<b>STEPS TO BE TAKEN IN CASE MATERIAL IS RELEASED OR SPILLED</b> Evacuate all personnel from affected area. Use appropriate protective equipment. If leak is in user's equipment, be certain to purge piping with an inert gas prior to attempting repairs. If leak is in container or container valve, contact your closest supplier location or call the emergency telephone number listed herein.
<b>WASTE DISPOSAL METHOD</b> Do not attempt to dispose of waste or unused quantities. Return in the shipping container <u>properly labeled, with any valve outlet plugs or caps secured and valve protection cap in place</u> to your supplier. For emergency disposal assistance, contact your closest supplier location or call the emergency telephone number listed herein.

## SPECIAL PROTECTION INFORMATION

<b>RESPIRATORY PROTECTION (Specify type)</b> Positive pressure air line with mask or self-contained breathing apparatus should be available for emergency use.	
<b>VENTILATION</b> Hood with forced ventilation	<b>SPECIAL</b> N/A
<b>MECHANICAL (Gen.)</b> N/A	<b>OTHER</b> N/A
<b>LOCAL EXHAUST</b> To prevent accumulation above the ceiling limit for HCl	
<b>PROTECTIVE GLOVES</b> Natural rubber, PVC or polyethylene (low density)	
<b>EYE PROTECTION</b> Safety goggles or glasses	
<b>OTHER PROTECTIVE EQUIPMENT</b> Safety shoes, safety shower, eyewash "fountain" face shield	

## SPECIAL PRECAUTIONS\*

<b>SPECIAL LABELING INFORMATION</b> DOT Shipping Name: Dichlorosilane DOT Shipping Label: Toxic Gas; Flammable Gas; Corrosive	DOT Hazard Class: Division 2.3 II I.D. No.: UN 2189
<b>SPECIAL HANDLING AND STORAGE RECOMMENDATIONS</b> Protect cylinders from physical damage. Store in cool, dry, well-ventilated area of noncombustible construction away from heavily trafficked areas and emergency exits. Do not allow the temperature where cylinders are stored to exceed 125F (52C). Valve protection caps must remain in place unless container is secured with valve outlet piped to use point. Do not drag, slide or roll cylinders. Use a suitable hand truck for cylinder movement. Use a pressure reducing regulator when connecting cylinder to lower pressure (<7S psig) piping or systems. Do not heat cylinder by any means to increase the discharge rate of product from the cylinder. Use a check valve or trap in the discharge line to prevent hazardous back flow into the cylinder. Use a "first in – first out" inventory system to prevent full cylinders being stored for excessive periods of time.	
<b>SPECIAL PACKAGING RECOMMENDATIONS</b> Any materials suitable for use with anhydrous hydrogen chloride may be used with dichlorosilane. Systems and equipment must be kept scrupulously dry.	
<b>OTHER RECOMMENDATIONS OR PRECAUTIONS</b> Earth-ground and bond all lines and equipment associated with the dichlorosilane system. Electrical equipment should be non-sparking or explosion proof. Compressed gas cylinders should not be refilled except by qualified producers of compressed gases. NEVER transport cylinders in trunks of vehicles, enclosed vans, truck cabs or in passenger compartments. Transport cylinders secured in open flatbed or in open pick-up type vehicles.	

Information contained in this material safety data sheet is offered without charge for use by technically qualified personnel at their discretion and risk. All statements, technical information and recommendations contained herein are based on tests and data which we believe to be reliable, but the accuracy or completeness thereof is not guaranteed no warranty of any kind is made with respect thereto. This information is not intended as a license to operate under or a recommendation to practice or infringe any patent of this Company or others covering any process, composition of matter or use.

Since the Company shall have no control of the use of the product described herein, the Company assumes no liability for loss or damage incurred from the proper or improper use of such product.